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CHAPTER . 11 . REVIEW . Gases . SHORT ANSWER . Answer the following questions in the space provided. 1. Consider the following data table:
Approximate pressure (kPa) Altitude above sea level (km) 100 . o (sea level) 50 5.5 (peak of Mt. Kilimanjaro) 25 11 Get cruising altitude} < 0.1 22 (ozone layer) ~.

REVIEW Gases

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Answer the following questions in the
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pressure kPa 100 R 50 25 < 0.1 Altitude
above sea level (km) 0 (sea level) 5.5
(peak of Mt. Kilimanjaro) 11 (jet cruising
altitude) 22 (ozone layer) a.

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Chapter 11 Test Review. multiple choice (25) definition & applications of pressure (also atmospheric) SI unit of force. definition & use of a barometer. standard temperature & pressure (STP) Definition of Dalton's law of partial pressures. Definitions & formulas for Boyle's, Charles', Gay-Lussac's, and combined gas laws

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Composition of Gases MIXED REVIEW
SHORT ANSWER Answer the following questions in the space provided. 1. c The average speed of a gas molecule is most directly related to the . (a) polarity of the molecule (b) pressure of the gas (c) temperature of the gas (d) number of moles in the sample

CHAPTER 11 REVIEW Molecular Composition of Gases

Review Previous Concepts Gases
CHAPTER 11 1. What is a mole? What information does the molar mass of a ...
344 CHAPTER 11. Dalton's Law of Partial Pressures The pressure exerted by each gas in a mixture is called the ... of 0.28 atm when mixed with one or more gases. Dalton's law of partial pressures states that the total

CHAPTER 11 Gases

states that the rates of effusion of gases at the same temperature and pressure are inversely proportional to the square roots of their molar masses. Graham's

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law of effusion the process whereby the molecules of a gas are confined in a container randomly pass through a tiny opening in the container

Study 60 Terms | Chapter 11: Gases- PAP Chemistry Review ...

Chapter 11 181 Chapter 11 - Gases
Review Skills 11.1 Gases and Their
Properties Ideal Gases Properties of
Gases Discovering the Relationships
Between Properties The Relationship
Between Volume ... neon gas mixed with
argon gas. (Objs 21 & 22) a. If the total
pressure of the mixture of gases is 1.30
kPa and the partial pressure of neon

Chapter 11 - Gases

Gases and Pressure In the chapter
"States of Matter," you read about the
kinetic- ... 344 CHaPTeR 11. The total
pressure of a gas mixture is the sum of
the pressures of the gases in it. ... of
0.28 atm when mixed with one or more
gases.

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SECTION 11.1 Gases and Pressure - Pickford High School

CHAPTER 11 REVIEW Gases SECTION 2
SHORT ANSWER Answer the following
questions in the space provided. 1. State
whether the pressure of a fixed mass of
gas will increase, decrease, or stay the
same in the following circumstances:
increase a. temperature increases,
volume stays the same decrease b.
volume increases, temperature stays the
same

mc06se cFMsR i-vi - Ed W. Clark High School

CHAPTER 10 REVIEW Physical
Characteristics of Gases. CHAPTER 10
REVIEW Physical Characteristics of
Gases MIXED REVIEW SHORT ANSWER
Answer the following... Filesize: 524 KB;
Language: English ... Gases and
Atmospheric Chemistry Chapter 11: ...
The Gas State and Gas Laws 11-5 5.
Answers may vary. Sample answer:
Table 1 Methanal products from which

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WS Section Assessment (#1-7) 1.1 Quiz
1.2 u2013 Chemistry Far and Wide
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